| REPORT DOCUMENTATION PAGE | | | Form Approved OMB NO. 0704-0188 |
|--|---|--|--|
| gathering and maintaining the data needed, ar collection of information, including suggestions | nd completing and reviewing the collection of infor | mation. Send comment regarding arters Services. Directorate for a | wing instructions, searching existing data sources, this burden estimates or any other aspect of this information Operations and Reports, 1215 Jefferson ect (0704-0188), Washington, OC 20503. |
| 1. AGENCY USE ONLY (Leave blank) | 2. REPORT DATE | 3. REPORT TYPE | AND DATES COVERED |
| | 1/22/97 | Technical | |
| Development of Microwave Ferrites for High Performance Applications | | | 5. FUNDING NUMBERS . DAAH04-95-1-0278 |
| 6. AUTHOR(S) | | | |
| Dr. Vasantha R.W. Amarakoon, Dr. Vernon L. Burdick, Yong S. Cho, Wesley Garbic | | | |
| 7. PERFORMING ORGANIZATION NAMES(S) AND ADDRESS(ES) New York State College of Ceramics at Alfred University 2 Pine Street | | | 8. PERFORMING ORGANIZATION REPORT NUMBER |
| Alfred, NY 14802-1296 | | | #13-75-86 |
| 9. SPONSORING / MONITORING AGENCY NAME(S) AND ADDRESS(ES) | | | 10. SPONSORING / MONITORING AGENCY REPORT NUMBER |
| U.S. Army Research Office P.O. Box 12211 Research Triangle Park, NC 27709-2211 | | | ARO 27548.15-MS |
| 11. SUPPLEMENTARY NOTES | | | |
| The views, opinions and/or findings contained in this report are those of the author(s) and should not be construed as an official Department of the Army position, policy or decision, unless so designated by other documentation. | | | |
| :2a. DISTRIBUTION / AVAILABILITY STATEMENT | | | 12 b. DISTRIBUTION CODE |
| Approved for public release; distribution unlimited. | | | |
| 13. ABSTRACT (Maximum 200 words) | | | |
| LiZn ferrites were investigated to control the resultant microstructure or to obtain nanoscale powders at a low temperature. Two nonconventional chemical methods are introduced. The sol-gel coating procedure was used to uniformly distribute additives like SiO ₂ and Bi ₂ O ₃ . These oxides improved the densification of the LiZn ferrites at low temperatures via a liquid phase sintering. Nanocrystalline LiZn ferrites were obtained at a low temperature of 400°C via the assistance of a combustible acid like polyacrylic acid. | | | |
| 19970210 119 | | | |
| 14 SUBJECT TERMS LiZn Ferrite, Sol-Gel, Nanoparticle, Microstructure | | | 15. NUMBER IF PAGES |
| | | | 16. PRICE CODE |
| 17 SECURITY CLASSIFICATION 18. S | SECURITY CLASSIFICATION 19. | SECURITY CLASSIFICA | ATION 20. LIMITATION OF ABSTRACT |
| | | OF ABSTRACT | |

DEVELOPMENT OF MICROWAVE FERRITES FOR HIGH PERFORMANCE APPLICATIONS

TECHNICAL REPORT

DR. VASANTHA R. W. AMARAKOON DR. VERNON L. BURDICK YONG S. CHO WESLEY GARBIC

1/22/97

U.S. ARMY RESEARCH OFFICE

NEW YORK STATE COLLEGE OF CERAMICS ALFRED UNIVERSITY

APPROVED FOR PUBLIC RELEASE; DISTRIBUTION LIMITED

THE VIEWS, OPINIONS, AND/OR FINDINGS CONTAINED IN THIS REPORT ARE THOSE OF THE AUTHORS AND SHOULD NOT BE CONSTRUED AS AN OFFICIAL DEPARTMENT OF THE ARMY POSITION, POLICY, OR DECISION. UNLESS SO DESIGNATED BY OTHER DOCUMENTATION

Development of Microwave Ferrites for High Performance Applications

Two chemical methods applied to LiZn ferrites will be briefly introduced. The influences of the experimental parameters and compositions on the final microstructures and properties are being investigated. Several reprints and manuscripts resulted from the previous research on yttrium iron garnet materials are enclosed (reported in technical report submitted to ARO on 2/5/96).

I. Microstructural Control of LiZn Ferrites with Chemically Derived Additives

Raw materials such as Li₂CO₃, ZnO and Fe₂O₃ were mixed using yttria stabilized zirconia balls (Tosoh Co.) and calcined at 800°C for 4 hrs to form a spinel phase. Two additives, Bi₂O₃ - MnO₂ and SiO₂ - MnO₂ were incorporated into Li_{0.3}Zn_{0.4}Fe_{2.3}O₄ via the sol-gel coating method. The calcined powders were milled using the same media. The experimental procedures for sol-gel coating are as follows. In case of Bi₂O₃ - MnO₂ coating, Bi nitrate and Mn acetate were used as raw materials. Bi nitrate pentahydrate, Bi(NO₃)₃·5H₂O and Mn acetate tetrahydrate, Mn(CH₃COO)₂·4H₂O corresponding to 1 wt% Bi₂O₃ and 1 wt% MnO₂ were dissolved completely into deionized water. A 50 wt% aqueous solution of polyacrylic acid (PAA) was added to the solution resulting in a gelatinous white precipitate. And then the calcined LiZn powder was mixed with the gel. For the SiO₂-MnO₂ coating, tetraethylorthosilicates (TEOS), Si(OC₂H₅)₄ and Mn acetate tetrahydrate, Mn(CH₃COO)₂·4H₂O corresponding to 1 wt% SiO₂ and 1 wt% MnO₂ were dissolved into ethanol. A few drops of HCl was added as a catalyst for sol-gel reaction. The calcined powder of LiZn ferrite was added into the clear solution and kept stirring for a few hours for complete reaction. All coated powders were dried at 70°C and then pulverized using a mortar. Pellets were made by uniaxial pressing of the powder at around 12,000 psi and sintered at 1050°C for 2 hrs with a heating rate of 300°C/hr after organic burnout at 500°C.

The preliminary observation of microstructures showed that Bi₂O₃ and SiO₂ enhanced grain growth and densification. Large grains were observed in these cases. Distinct grain boundary regions were seen after chemical etching in case of Bi₂O₃, indicating a weak grain boundary layer. While, in case of SiO₂, it is likely that the addition SiO₂ induces a glassy phase in the grain boundaries (not easily etchable). Based on the grain size variations with temperature and time, grain growth kinetics will be studied for the different compositions. Magnetic and dielectric properties in the microwave frequency

range will be also investigated. Additionally, different contents of the additives will be adjusted in order to optimize the magnetic and dielectric properties, .

II. Preparation of Nanocrystalline LiZn ferrites Using Polyacrylic Acid

A new chemical method utilizing a combustible polyacrylic acid (PAA) is introduced to prepare nanoscale ferrite powders. This method is relatively simple compared to other chemical methods. Nitrates of Li, Zn and Fe were dissolved in deionized water to form a 0.5 M solution. The nitrate solution was inserted into a 50 wt% polyacrylic acid aqueous solution. The atomic ratio of the carboxyl ion to cation was 0.5. After a few minutes, the clear nitrate solution changed into an opaque viscous gel having brown color. The gel was dried at 70°C and calcined from 400°C to 700°C for 30 min.

As a result of TEM observation, the average particle size of the LiZn ferrites fired at 400°C was around 10 nm. The XRD pattern indicated the crystalline LiZn ferrite phase. As the synthesis temperature increases, the particle size tended to increase but the nanoscale was kept until 600°C. From the initial results, it can be concluded that the PAA method is promising for the preparation of nanoscale LiZn ferrite powders at low temperatures. Densification behavior of the nanosized ferrites will be investigated.

List of Publications

- Y. S. Cho, V. L. Burdick and V. R. W. Amarakoon, "Hydrothermal Preparation and Morphology Characteristics of Y₃Fe₅O₁₂," *J. Am. Ceram. Soc.* (submitted, Paper No. 191632C)).
- Y. S. Cho, D. T. Hoelzer, L. Brissette and V. R. W. Amarakoon, "Grain Boundary and Electrical Characteristics of Chemically Modified Gd-YIG," *J. Mater. Res.* (submitted, Paper No. 96-271).
- Y. S. Cho and V. R. W. Amarakoon, "Nanoscale Coating of Silicon and Manganese on Ferrimagnetic Yttrium Iron Garnets," *J. Am. Ceram. Soc.*, **79**(10) 2755-58 (1996).
- Y. S. Cho, L. Brissette and V. R. W. Amarakoon, "Effect of Sol-Gel Coating on the Microstructure and Electrical Properties of Microwave Garnets," *Mater. Res. Soc. Symp. Proc.*, Vol. 411, Electrically Based Microstructural Characterization, pp287-292 (1996).

List of Presentation

- Preparation of Nanocrystalline Ferrites via the Assistance of Polyacrylic Acid,, *The 99th Annual Meeting of Am Ceram. Soc.*, Cincinnati, OH, May 1997.
- "Grain Growth and Microstructural Characteristics of LiZn Ferrites with Chemically Derived Additives," *The Fall Meeting of Mater. Res. Soc.*, Boston, MA, Dec. 1996.
- Hydrothermal Synthesis of Ferrimagnetic Yttrium Iron Garnets, *The 98h Annual Meeting of Am Ceram. Soc.*, Indianapolis, IN, Apr. 1996.
- Effect of Sol-Gel Coating on the Microstructure and Properties of Microwave Garnets, The Fall Meeting of Mater. Res. Soc., Boston, MA, Nov. 1995.
- Microstructure and Phase Evolution in Ferrimagnetic Garnets Prepared with Additives via Sol-Gel Coating, *The 97th Annual Meeting of Am Ceram. Soc.*, Cincinnati, OH, May 1995.